

Contacts: eveline.mayner@epfl.ch
manfred.zinn@epfl.ch



Biochemical Engineering

Exercise Session 7 Correction

1) Chemostat Theory

Draw a x-D diagram with consideration of maintenance energy and insert, biomass, volumetric productivity, specific productivity, residual glucose concentration s , D_{opt} and D_{crit} .

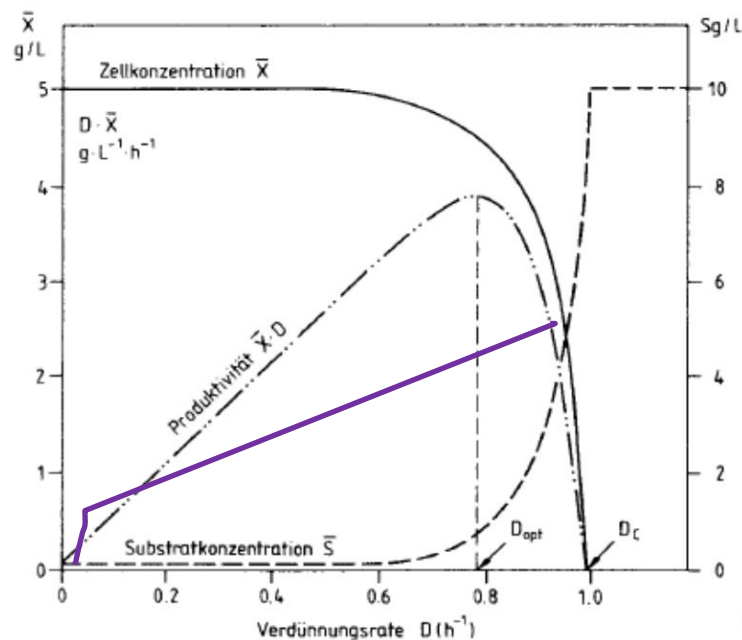


Figure 1: The purple function corresponds to the specific productivity $q_p = \mu/Y_{x/s}$

What do you understand under a turbidostat and a pH auxostat?

In other techniques, a fermenter variable, e.g., turbidity or pH, will be monitored using an appropriate detector and the liquid flow rate will be automatically adjusted so as to maintain the variable at a constant level (**closed loop control**)

What are their differences to a chemostat?

The dilution rate is not constant, the specific growth rate of the cells is close to μ_{max} . Tricky to run in a stable way (fast analytics needed and good algorithms).

2) Substrate conversion and biomass productivity

A 5 m³ fermenter is operated continuously with a feed substrate concentration of 20 kg m⁻³. The genetically engineered *E. coli* cultivated in the reactor has the following characteristics:

$$\mu_{\max} = 0.45 \text{ h}^{-1}; K_s = 800 \text{ g m}^{-3}; Y_{X/S} = 0.55 \text{ kg kg}^{-1}$$

a) What feed flow rate is required to achieve 90% substrate conversion?

At steady state the substrate concentration is expressed as follows, with $S = 0.9 \cdot S_0$

$$S = K_s \frac{D}{\mu_{\max} - D} \Leftrightarrow D = \frac{S \cdot \mu_{\max}}{K_s + S} = \frac{2 \frac{\text{kg}}{\text{m}^3} \cdot 0.45 \frac{1}{\text{h}}}{0.8 \frac{\text{kg}}{\text{m}^3} + 2 \frac{\text{kg}}{\text{m}^3}} = 0.321 \frac{1}{\text{h}}$$

$$Q = 0.321 \frac{1}{\text{h}} \cdot 5 \text{ m}^3 = 1.6 \frac{\text{m}^3}{\text{h}}$$

b) How does the biomass productivity at 90% substrate conversion compare with the maximum possible?

$$D_m = D_{\text{opt}} = \mu_{\max} \left(1 - \sqrt{\frac{K_s}{K_s + S_0}} \right) = 0.45 \frac{1}{\text{h}} \left(1 - \sqrt{\frac{0.8 \frac{\text{kg}}{\text{m}^3}}{0.8 \frac{\text{kg}}{\text{m}^3} + 20 \frac{\text{kg}}{\text{m}^3}}} \right) = 0.362 \frac{1}{\text{h}}$$

c) What is the biomass concentration in case of an at the optimal dilution rate?

$$X = Y_{X/S} \left(S_0 - \frac{K_s \cdot D}{\mu_{\max} - D} \right) = 0.55 \left(20 - \frac{0.8 \cdot 0.321}{0.45 - 0.321} \right) = 9.9 \frac{\text{kg}}{\text{m}^3}$$

$$X_{\text{opt}} = Y_{X/S} \left(S_0 - \frac{K_s \cdot D_{\text{opt}}}{\mu_{\max} - D_{\text{opt}}} \right) = 0.55 \left(20 - \frac{0.8 \cdot 0.362}{0.45 - 0.362} \right) = 9.19 \frac{\text{kg}}{\text{m}^3}$$

3) Growth inhibition

The specific growth rate for inhibited growth in a chemostat is given by the following equation:

$$\mu = \mu_{\max} S / (K_s + S + I K_s / K_i)$$

Where

$$s_0 = 10 \text{ g L}^{-1}, K_s = 1 \text{ g L}^{-1}; I = 0.05 \text{ g L}^{-1}, Y_{X/S} = 0.1 \text{ g g}^{-1}$$

$$x_0 = 0, K_i = 0.01 \text{ g L}^{-1}, \mu_{\max} = 0.5 \text{ h}^{-1}$$

a) Determine x and s as function of D when $I = 0$

The equation of the growth rate can be rearranged and solved for the substrate concentration,

$$S = K_s \frac{D}{\mu_{\max} - D} \quad \text{if } I = 0 \text{ and } \mu = D$$

The same result is also obtained if the general formula expressing the biomass concentration is used

$$X = Y_{X/S}(S_0 - S) = Y_{X/S} \left(S_0 - K_S \frac{D}{\mu_{max} - D} \right) \Leftrightarrow S = K_S \frac{D}{\mu_{max} - D}$$

- b) With inhibitor added to a chemostat, determine the effluent substrate concentration and x as function of D

$$S = \frac{D(K_S + \frac{IK_S}{K_I})}{\mu_{max} - D}$$

$$X = Y_{X/S} \left(S_0 - \frac{\left(\left(K_S + \frac{IK_S}{K_I} \right) \cdot D \right)}{\mu_{max} - D} \right)$$

- c) Determine the volumetric cell productivity, DX , as a function of dilution rate

The productivity is expressed by $P_X = D \cdot X$ with the biomass concentration X calculated in part b.

4) Wash-out experiment

What is the biomass concentration in a chemostat when one knows the following parameters:

$$V = 2.200 \text{ L} \quad F = 200 \text{ mL/h}$$

$$x_0 = 10 \text{ g L}^{-1} \text{ (at } t = 0 \text{ h)} \quad \mu_{max} = 0.3 \text{ h}^{-1}$$

$$K_S = 0.2 \text{ g glucose/L} \quad Y_{X/S} = 0.5 \text{ g cells / g glucose}$$

(consider not metabolized glucose s!)

What is the concentration of the biomass at different times when the dilution rate was changed to $D = 1.0 \text{ h}^{-1}$? Fill in the table below.

What is the concentration of the biomass at different times?

$$\ln X = (\mu_{max} - D) \cdot t + \ln X_0$$

$$X = 1000 \cdot X_0 \exp((\mu_{max} - D) \cdot t) \text{ (in mg/L)}$$

Time [h]	Biomass concentration [mg L ⁻¹] (3 digits after the comma !)
1	2461.336
5	149.6764
15	0.136
25	0.000